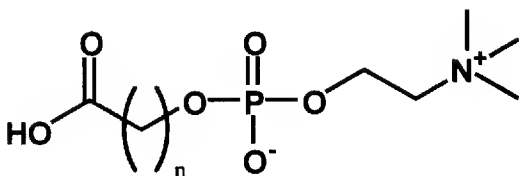


IN THE CLAIMS:

Kindly rewrite Claims 1 and 2 as follows:

1. (Currently Amended) A method of ~~manufacturing after treating~~ an eye lens material which prevents protein adsorption comprising in which a phosphorylcholine group containing chemical compound is reacted reacting and covalently bonding through ester-bonding to the eye lens material, in an organic solvent, carboxymethyl phosphorylcholine bonded in an after treatment onto the surface of an eye lens material having hydroxyl groups, wherein the chemical compound represented by the following formula (2); is reacted and covalently bonded through ester bonding to the eye lens material in an organic solvent

[Chemical formula 2]



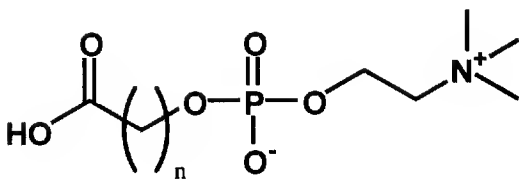
~~in which~~ wherein $n=1$, onto the surface of an eye lens material having hydroxyl groups, said carboxymethyl phosphorylcholine obtained by the oxidative cleavage of 1- α -glycerophosphorylcholine using periodate and ruthenium trichloride in a water/acetonitrile mixed solvent denotes a natural number 1-18.

2. (Currently Amended) A method of ~~manufacturing after treating~~ an eye lens material which prevents adsorption comprising: in which a phosphorylcholine group containing chemical compound is reacted and covalently bonded in an after treatment to the surface of an eye lens material, wherein hydroxyl groups are introduced

(a) introducing hydroxyl groups to said eye lens material by means of a plasma treatment, and then ~~the chemical compound represented by the following formula (2) is reacted~~

(b) reacting and covalently bonding through ester-bonding to the eye lens material, in an organic solvent, carboxymethyl phosphorylcholine represented by the following formula (2), bonded through ester-bonding to the eye lens material in an organic solvent;

[Chemical formula 2]



wherein in which n=1, onto the surface of an eye lens material having hydroxyl groups, said carboxymethyl phosphorylcholine obtained by the oxidative cleavage of 1- α -glycerophosphorylcholine using periodate and ruthenium trichloride in a water/acetonitrile mixed solvent denotes a natural number 1-18.

3 - 6. (Cancelled)